

4-Methyl-3,5-dinitrobenzoic acid–dimethyl sulfoxide (1/1)

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Key indicators

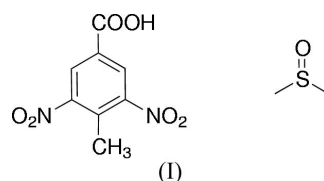
Single-crystal X-ray study
 $T = 180\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.041
 wR factor = 0.102
Data-to-parameter ratio = 15.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title complex, $\text{C}_8\text{H}_6\text{N}_2\text{O}_6 \cdot \text{C}_2\text{H}_6\text{OS}$, was predicted to illustrate an intermolecular hydrogen-bond motif between the carboxylic acid and the sulfoxide functionalities, based upon a previously published structure of an analogous complex. The predicted hydrogen-bond motif was observed, thereby indicating a certain robustness of this intermolecular interaction for crystal engineering purposes.

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Comment

The asymmetric unit of the title crystal structure, (I), consists of one molecule each of 4-methyl-3,5-dinitrobenzoic acid and dimethyl sulfoxide (DMSO) (Fig. 1).



The crystallization was performed to evaluate the robustness of an intermolecular hydrogen bond involving an $\text{O}-\text{H} \cdots \text{O}=\text{S}$ contact between a carboxylic acid and a sulfoxide. This interaction was recently observed in the crystal structure of an analogous complex involving 3,5-dinitrobenzoic acid and DMSO (Abthorpe *et al.*, 2005). This interaction also is found in 29 of a possible 37 instances in the Cambridge Structural Database (CSD Version 5.25 Update 3; Allen, 2002), when searching for structures which contain both a carboxyl group and a DMSO molecule among all organic structures for which three-dimensional coordinates have been determined. The hydrogen-bond interaction in the crystal structure is presented in Fig. 2.

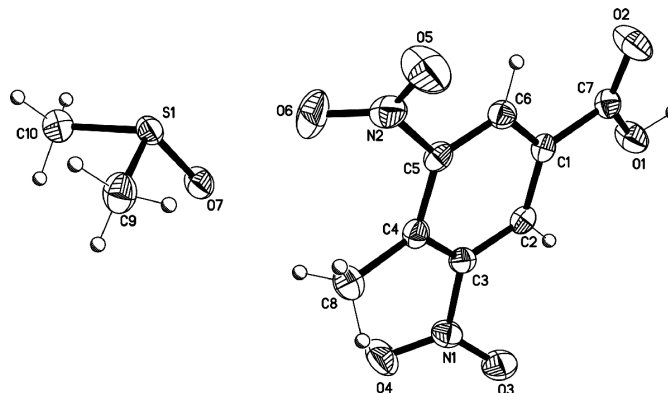


Figure 1
The asymmetric unit (*XP*; Sheldrick, 1993) of (I), showing displacement ellipsoids at the 50% probability level.

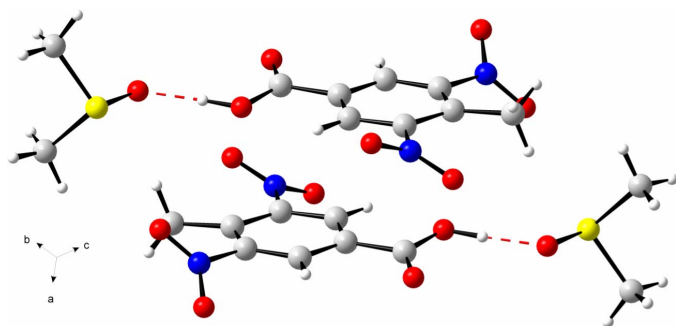


Figure 2
Part of the crystal packing (*DIAMOND*; Brandenburg, 1999), showing intermolecular hydrogen-bond interactions as dashed lines.

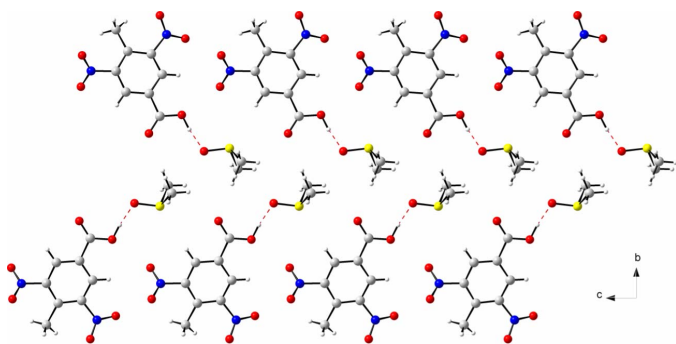


Figure 3
The crystal packing (*DIAMOND*; Brandenburg, 1999), viewed along [100], showing sheets stacking along [010].

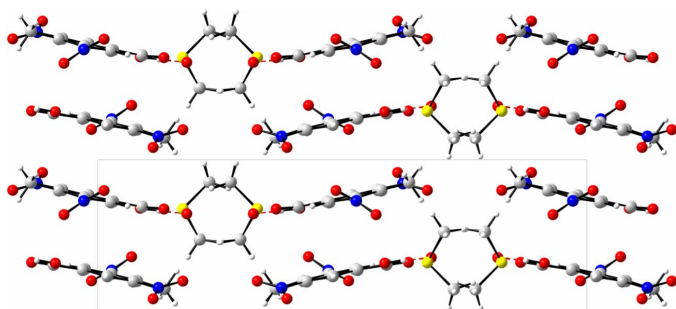


Figure 4
The crystal packing (*DIAMOND*; Brandenburg, 1999), viewed along [001], showing sheets stacking along [010].

The title complex packs in a monoclinic unit cell in the space group $P2_1/c$. Crystal packing results in alternating sheets of acid and DMSO molecules stacking along [010]. (Figs. 3 and 4).

The experiment reported here represents a successful demonstration of the methodological approach of crystal engineering: observation of a particular heteromolecular hydrogen-bonding interaction, evaluation of the abundance of the interaction in the CSD, and application of this information to the design of a novel crystalline molecular complex. The demonstrated robustness of this hydrogen-bond motif indicates a potential utility for future crystal engineering experiment design.

Experimental

All starting components were obtained from Sigma Aldrich Ltd. 4-Methyl-3,5-dinitrobenzoic acid (64 mg) was dissolved in excess DMSO with gentle heating. The resulting solution was allowed to cool and evaporate slowly over a period of one week. From the solids that precipitated, a single crystal was harvested for subsequent XRD analysis.

Crystal data

$C_8H_6N_2O_6 \cdot C_2H_6OS$
 $M_r = 304.28$
 Monoclinic, $P2_1/c$
 $a = 6.9483$ (2) Å
 $b = 22.4844$ (5) Å
 $c = 8.2364$ (2) Å
 $\beta = 92.765$ (1)°
 $V = 1285.26$ (6) Å³
 $Z = 4$

$D_x = 1.572$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 7863 reflections
 $\theta = 1.0$ – 27.5 °
 $\mu = 0.29$ mm⁻¹
 $T = 180$ (2) K
 Plate, colourless
 $0.35 \times 0.32 \times 0.10$ mm

Data collection

Nonius KappaCCD diffractometer
 Thin-slice ω and φ scans
 Absorption correction: multi-scan
 (SORTAV; Blessing, 1995)
 $T_{\min} = 0.901$, $T_{\max} = 0.976$
 9790 measured reflections
 2930 independent reflections

2263 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.043$
 $\theta_{\max} = 27.5$ °
 $h = -9 \rightarrow 9$
 $k = -29 \rightarrow 29$
 $l = -7 \rightarrow 10$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.102$
 $S = 1.06$
 2930 reflections
 187 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0398P)^2 + 0.6419P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.24$ e Å⁻³
 $\Delta\rho_{\min} = -0.33$ e Å⁻³

All H atoms bonded to carbon were positioned geometrically and refined using a riding model, with $U_{\text{iso}} = 1.5U_{\text{eq}}$ for methyl H atoms and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier atom})$ for all other H atoms. The C–H distances of the methyl groups were fixed at 0.98 Å; all other C–H distances were fixed at 0.95 Å. The O–H H atom was located in a difference Fourier map and refined isotropically.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *HKL DENZO* (Otwinowski & Minor, 1997) and *SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Sheldrick, 1993) and *DIAMOND* (Brandenburg, 1999)(software used to prepare material for publication: *SHELXL97*).

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